

SYNTHESIS, SPECTROSCOPIC CHARACTERIZATION AND *IN VITRO* ANTIMICROBIAL EVALUATION OF NOVEL PYRROLO BENZIMIDAZOLE DERIVATIVES

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ABSTRACT

A series of Pyrrolo Benzimidazole derivatives were synthesized by reaction of 4'-cyano-3'-oxo-2'-dihydro-1H-pyrrolo [1,2-a] benzimidazole **1(a-b)** with chloroacetyl chloride formed *N*-chloroacetyl compounds **2(a-b)** which on further treated with thiourea affords 2-aminothiazole compounds **3(a-b)** which on treatment with methyl iodide in presence of tri-ethyl amine gave *N,N*-dimethyl-aminothiazole compounds **4(a-b)** which were treated with Hydrochloric acid formed 4-amide derivatives of pyrrolobenzimidazole **5(a-b)**. **4(a-b)** treated with sodium azide and ammonium chloride in DMF formed tetrazole compounds **6(a-b)**. All synthesized novel derivatives were characterized by Spectral and Elemental analysis. These benzimidazole derivatives were tested against fungi, Gram-positive and Gram-negative bacteria. These compounds showed remarkable antimicrobial and fungicidal activity.

Keywords: 2-aminothiazole, antimicrobial activity, fungicidal activity, pyrrolobenzimidazole.

INTRODUCTION

Benzimidazoles are biologically important molecules and they are very useful for the synthesis of pharmaceutically active compounds. 1 or 2 position substituted benzimidazoles derivatives shows wide-ranging therapeutic applications¹⁻⁷. Aminothiazole commonly used as precursor for the synthesis of many active compounds including sulpha drugs, biocides, fungicides. 2-Aminothiazole has antibacterial activity⁸ and also used as a therapeutic drug for prion diseases⁹. Amide derivatives exhibit variety of biological activities like antibacterial¹⁰⁻¹¹, antifungal¹². Tetrazole derivatives exhibit different biological activities such as antimicrobial¹³, antiviral¹⁴, antiallergic¹⁵ and anti-inflammatory¹⁶. When aminothiazole, amides and tetrazole are conjugates with other heterocyclic compound, enhance biological activity. Hence during extensive work on

synthesis of Benzimidazole analogs, substituted aminothiazole, amide and tetrazole with pyrrolobenzimidazoles were synthesized.

MATERIALS AND METHODS

The reactions were monitored by TLC using 0.25 mm E-Merck silica gel plates, which were visualized in Iodine Chamber. Melting points were taken in open capillaries and are uncorrected. ¹H NMR spectra was performed in DMSO-d₆ on 300 MHz using TMS as an internal standard.

Synthesis of Compounds

4'-Cyano-2'-dihydro-3'-oxo-3-(2-chloroacetyl)-1H-pyrrolo[1,2-a] benzimidazole **2(a-b)**

Triethyl amine (6.20g, 0.060 mole) was added to the solution of 4'-cyano-3'-oxo-2'-dihydro-1H-pyrrolo [1,2-a]benzimidazole **1(a-b)** (10.0g,

0.050 mole) in dry benzene (50 ml) and stirred for 30min at room temperature. The reaction mass was chilled at 0-5^o C then chloroacetyl chloride was added (6.0g, 0.053 mole) under stirring below 5^o C. The reaction mass was subsequently refluxed for 4 h on water bath. The completion of the reaction was monitored by TLC using n-hexane: ethyl acetate (7:3) as a solvent system. The solvent was removed under reduced pressure and the residue was poured into water. The solid obtained was recrystallized by methanol to get pure product 4'-cyano-2'-dihydro-3'-oxo-3-(2-chloroacetyl)-1H-pyrrolo[1,2-a]benzimidazole **2(a-b)**.

4'-Cyano-2'-dihydro-3'-oxo-3-(2''-amino-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole 3(a-b)

Brown powder of 4'-cyano-2'-dihydro-3'-oxo-3-(2-chloroacetyl)-1H-pyrrolo[1,2-a]benzimidazole **2(a-b)** (5.5g, 0.02 mole) was added in dry acetone (50 ml) and stirred for 10 min. Then thiourea (1.53 g, 0.02 mole) was added and refluxed for 8 h. The completion of the reaction was monitored by TLC using n-hexane: ethyl acetate (7:3) as solvent system. After completion of reaction excess of acetone was distilled off and the residue obtained was poured into crushed ice which was filtered, dried and recrystallized from methanol which gave 4'-cyano-2'-dihydro-3'-oxo-3-(2''-amino-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **3(a-b)**.

4'-Cyano-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole 4(a-b)

To the solution of 4'-cyano-2'-dihydro-3'-oxo-3-(2''-amino-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **3(a-b)** (10.0g, 0.033 mole) in DMF (10 ml) triethyl amine (4.2 g, 0.041 mole) was added at room temperature. Then methyl iodide (10.0 g, 0.070 mole) was added into the reaction mixture and stirred at room temperature. The completion of the reaction was monitored by TLC using n-hexane: ethyl acetate (5:5) as a solvent system. The reaction was accomplished within 5 h at room temperature. After accomplished reaction water was added in the reaction mass resulting into brown coloured precipitate which was separated by filtration. The crude solid was recrystallized from ethanol/water (4:1) to get pure product of 4'-cyano-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **4(a-b)**.

4'-Carbamoyl-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole 5(a-b)

4'-cyano-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **4(a-b)** (10 g, 0.031 mole) and conc. HCl (40 ml) was immersed in the flask at 40^o C. The reaction mixture was stirred vigorously at 40^o C for 3 h. The progress and completion of the reaction was monitored by TLC using n-hexane: ethyl acetate (5:5) as a solvent system. After completion of reaction, reaction mass was poured onto ice-water afford 4'-carbamoyl-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **5(a-b)**. The obtained solid was filtered off, washed with cold water, dried and recrystallized from ethanol.

3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-2'-dihydro-4'-[(1''-H(1''',2''',3''',4''')) tetrazol]-3'-oxo-1H-pyrrolo[1,2-a]benzimidazole 6(a-b)

Process for preparation of tetrazoles was as per US 5744612 A Ref. ¹⁷

Sodium azide (3.0 g, 0.046 mole) and ammonium chloride (2.50 g, 0.046 mole) was stirred at room temperature in 50 ml xylene. To this reaction mixture powder of 4'-cyano-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole **4(a-b)** (10.0 g, 0.031 mole) was added at room temperature, then the reaction mixture was heated with stirring at 125 to 130^oC. The progress and completion of reaction was monitored by TLC using n-hexane: ethyl acetate (5:5) as a solvent system. After completion of the reaction, the reaction mixture was cooled to 100^o C. After the solvent under reduced pressure was distilled out, oily mass was obtained. To this, 50 ml water and 36 % hydrochloric acid (5.0g, 0.048 mole) were added, thus precipitate was obtained. The precipitate was filtered, dried and recrystallized from ethanol gave 3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-2'-dihydro-4'-[(1''-H(1''',2''',3''',4'''))tetrazol]-3'-oxo-1H-pyrrolo[1,2-a]benzimidazole **6(a-b)**.

Characterization of Synthesized Compounds

4'-Cyano-2'-dihydro-3'-oxo-3-(2-chloroacetyl)-1H-pyrrolo[1,2-a]benzimidazole (2a)

Molecular Formula: C₁₃H₈ClN₃O₂, Melting Point: 201-204^oC, Yield: 65 %, Elemental Analysis % (Calculated) Found: C (57.09) 57.19, H (2.95) 3.01, N (15.35) 15.51. IR (KBr): 2996 (CH), 2245 (CN), 1747 (>C=O), 1650 (HN>C=O), 1559, 1203, 1060,760 cm⁻¹. ¹H NMR (DMSO-d₆): δ 3.63 (s, 2H, CH₂), δ 4.06 (s, 2H, CH₂), δ 6.64-6.90 (m, 4H, Ar-H). M⁺²: 273.67.

4'-cyano-2'-dihydro-3'-oxo-5-fluoro-3-(2-chloroacetyl)-1H-pyrrolo[1,2-a]benzimidazole (2b)

Molecular Formula: C₁₃H₈ClFN₃O₂, Melting Point: 204-208°C, Yield: 65%, Elemental Analysis % (Calculated) Found: C (53.53) 53.45, H (2.42) 2.57, Cl (12.16) 12.23, F (4.73) 4.77, N (14.41) 14.31. IR (KBr): 2966 (CH), 2258 (CN), 1730 (>C=O), 1660 (HN>C=O), 1556, 1213, 1040, 765 cm⁻¹. ¹H NMR (DMSO-d₆): δ 3.65 (s, 2H, CH₂), δ 4.16 (s, 2H, CH₂), δ 6.69-6.90 (m, 3H, Ar-H). M⁺: 291.66.

4'-cyano-2'-dihydro-3'-oxo-3-(2''-amino-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (3a)

Molecular Formula: C₁₄H₉N₅OS, Melting Point: 188-192°C, Yield: 55%, Elemental Analysis % (Calculated) Found: C (56.94) 57.05, H (3.07) 3.29, N (23.71) 23.61, S (10.86) 10.65. IR (KBr): 3512 (NH₂), 3040 (CH), 2257 (CN), 1760 (>C=O), 1550, 1185, 1079 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): at δ 3.88 (s, 2H, CH₂), δ 5.22 (s, 1H, CH), δ 6.60-6.90 (m, 4H, Ar-H), δ 5.85 (s, 2H, NH₂, D₂O exchangeable). M⁺: 295.37.

4'-cyano-2'-dihydro-3'-oxo-5-fluoro-3-(2''-amino-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (3b)

Molecular Formula: C₁₄H₈FN₅OS, Melting Point: 276-277°C, Yield: 55%, Elemental Analysis % (Calculated) Found: C (53.67) 53.55, H (2.57) 2.47, N (22.35) 22.21, F (6.06) 6.15, S (10.23) 10.23. IR (KBr): 3440 (NH₂), 2980 (CH), 2235 (CN), 1737 (>C=O), 1540, 1175, 1069 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): at δ 3.93 (s, 2H, CH₂), δ 5.19 (s, 1H, CH), δ 6.70-6.80 (m, 3H, Ar-H), δ 5.65 (s, 2H, NH₂, D₂O exchangeable). M⁺: 313.33.

4'-cyano-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (4a)

Molecular Formula: C₁₆H₁₅N₅OS, Melting Point: 193-198°C, Yield: 55%, Elemental Analysis % (Calculated) Found: C (59.43) 59.33, H (4.02) 4.09, N (21.66) 21.51, S (9.92) 10.01. IR (KBr): 3022 (CH), 2241 (CN), 1757 (>C=O), 1556, 1185, 1071 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): at δ 2.78 (s, 6H, N-CH₃), δ 3.90 (s, 2H, CH₂), δ 5.45 (s, 1H, CH), δ 6.45-6.90 (m, 4H, Ar-H). M⁺: 323.37.

4'-cyano-2'-dihydro-3'-oxo-5-fluoro-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (4b)

Molecular Formula: C₁₆H₁₅N₅FOS, Melting Point: 188-192°C, Yield: 55%, Elemental Analysis % (Calculated) Found: C (56.3) 56.55, H (3.54) 3.45, N (20.52) 20.52, F (5.57)

5.62, S (9.39) 9.52. IR (KBr): 3012 (CH), 2232 (CN), 1747 (>C=O), 1545, 1188, 1062 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): at δ 2.90 (s, 6H, N-CH₃), δ 3.98 (s, 2H, CH₂), δ 5.59 (s, 1H, CH), δ 6.36-7.76 (m, 3H, Ar-H). M⁺: 341.62.

4'-Carbamoyl-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (5a)

Molecular Formula: C₁₆H₁₅N₅O₂S, Melting Point: 180-184°C, Yield: 62%, Elemental Analysis % (Calculated) Found: C (56.29) 56.01, H (4.43) 4.39, N (20.51) 20.31, S (9.39) 9.28. IR (KBr): 3446 (NH₂), 2891 (CH), 1759 (>C=O), 1640 (H₂NC=O), 1575, 1362, 1198 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): δ 3.28 (s, 3H, N-CH₃), δ 3.89 (s, 2H, CH₂), δ 4.20 (s, 2H, NH₂, D₂O exchangeable), δ 4.60 (s, 1H, CH), δ 6.65-6.95 (m, 4H, Ar-H). M⁺: 341.39.

4'-Carbamoyl-2'-dihydro-3'-oxo-3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-1H-pyrrolo[1,2-a]benzimidazole (5b)

Molecular Formula: C₁₆H₁₄N₅FO₂S, Melting Point: 175-178°C, Yield: 67%, Elemental Analysis % (Calculated) Found: C (53.47) 53.01, H (3.93) 4.09, F (5.29) 5.39, N (19.49) 19.31, S (9.39) 9.28. IR (KBr): 3460 (NH₂), 2972 (CH), 1745 (>C=O), 1665 (H₂NC=O), 1560, 1316, 1184 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): δ 3.09 (s, 3H, N-CH₃), δ 3.89 (s, 2H, CH₂), δ 4.28 (s, 2H, NH₂, D₂O exchangeable), δ 4.75 (s, 1H, CH), δ 6.60-6.82 (m, 3H, Ar-H). M⁺: 359.39.

3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-2'-dihydro-4'-[(1''-H(1''',2''',3''',4'''))tetrazol]-3'-oxo-1H-pyrrolo[1,2-a]benzimidazole (6a)

Molecular Formula: C₁₆H₁₄N₈OS, Melting Point: 210-214°C, Yield: 56 %, Elemental Analysis % (Calculated) Found: C (52.45) 52.21, H (3.85) 3.96, N (30.58) 30.28, S (8.75) 8.68. IR (KBr): 3365 (NH), 3001 (CH), 1751 (>C=O), 1583, 1328, 1181 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): δ 3.21 (s, 3H, N-CH₃), δ 3.83 (s, 2H, CH₂), δ 5.19 (s, 1H, CH), δ 6.65-6.93 (m, 4H, Ar-H) and δ 8.48 (s, 1H, NH, D₂O exchangeable). M⁺: 366.40.

3-(2''-(dimethylamino)-1'',3''-thiazol-4'-yl)-6-Fluoro-2'-dihydro-4'-[(1''-H(1''',2''',3''',4'''))tetrazol]-3'-oxo-1H-pyrrolo[1,2-a]benzimidazole (6b)

Molecular Formula: C₁₆H₁₃FN₈OS, Melting Point: 195-198°C, Yield: 57%, Elemental Analysis % (Calculated) Found: C (49.99) 50.01, H (3.41) 3.67, F (4.94) 5.09, N (29.15) 30.31, S (8.34) 8.21. IR (KBr): 3350 (NH), 2981 (CH), 1726 (>C=O), 1561, 1323, 1156 cm⁻¹. ¹H NMR (DMSO-d₆/TMS): δ 3.10 (s, 3H, N-CH₃), δ 3.89 (s, 2H, CH₂), δ 4.99 (s, 1H, CH), δ 6.55-

6.73 (m, 3H, Ar-H) and δ 8.32 (s, 1H, NH, D₂O exchangeable). M⁺:384.39.

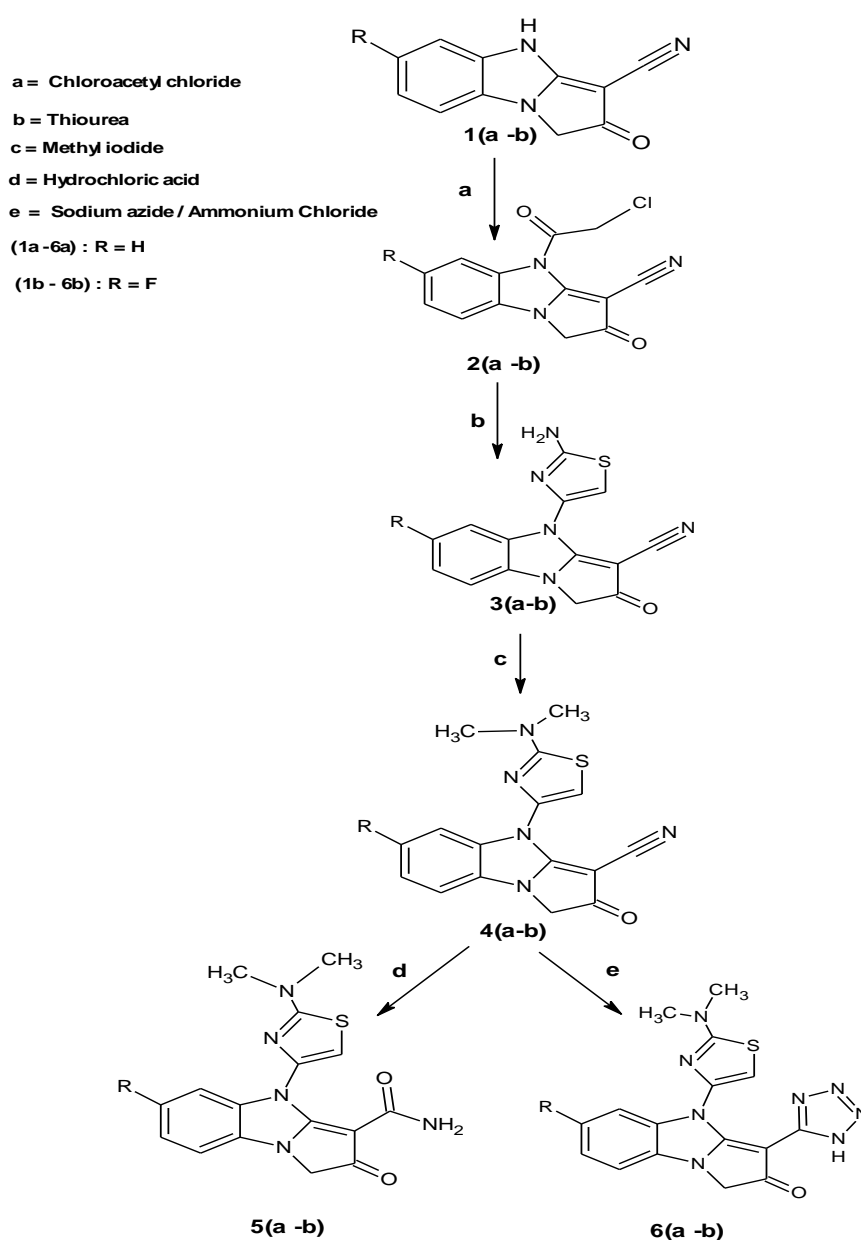
Antibacterial activity

Antibacterial activity was evaluated by the paper disc method. The Muller-Hinton agar and 5 mm diameter paper discs of whatman paper no. 1 were used. The compounds were dissolved in DMSO. The filter paper discs were soaked in different concentrations (50 μ g/ml and 100 μ g/ml) of the compounds, dried and then placed in the petriplates previously seeded with the test organisms *Salmonella typhi*, *Escherichia coli* and *Staphylococcus aureus*. The plates were incubated for 24–30 h at 28 \pm 2°C and the inhibition zone around each

disc was measured. The results are shown in Table: 1

Antifungal screening

The antifungal activity of the compounds was evaluated against *Aspergillus niger* by the paper disc method. The Sabouraud dextrose agar and 5 mm diameter paper discs of Whatman paper no.1 were used. The compounds were dissolved in DMSO. The filter paper discs were soaked in different solutions of the compounds, dried and then placed in the petriplates previously seeded with the test organisms *A. niger*. The plates were incubated for 48 h at 25 \pm 2°C and the inhibition zone around each disc was measured. The results are shown in Table: 1.



RESULTS AND DISCUSSION

In order to synthesize various pyrrolobenzimidazole derivatives, previously synthesized 4'-cyano-3'-oxo-2'-dihydro-1H-pyrrolo [1,2-a] benzimidazole **1(a-b)** was used as an important scaffold. **1(a-b)** reacted with chloroacetyl chloride formed *N*-acetyl chloride **2(a-b)**. IR spectra of compounds shows presence of strong band at 1747 cm^{-1} ($>\text{C}=\text{O}$) of amide, 1650 cm^{-1} ($>\text{C}=\text{O}$) of ketone and absence band at 3335 cm^{-1} of (NH). ^1HMR Spectra of compounds shows presence singlet's at δ 3.63 (s, 2H, CH_2), δ 4.06 (s, 2H, CH_2) and absence of singlet at δ 4.7 of NH. MS (m/z): 275.67 (M^{+2}). (Found: C, 57.19; H, 3.01; N, 15.51. $\text{C}_{13}\text{H}_8\text{ClN}_3\text{O}_2$ require C, 57.09; H, 2.95; N, 15.35). Thus the data obtained confirmed conversion of NH to *N*-acetyl chloride is achieved. **2(a-b)** which were treated with thiourea affords 2-aminothiozole derivatives **3(a-b)**. IR spectra of compounds shows presence of strong band at 3512 cm^{-1} (NH_2), 2257 cm^{-1} (CN), 1714 cm^{-1} ($\text{C}=\text{O}$) of ketone and absence of band at 1650 cm^{-1} ($\text{HN}-\text{C}=\text{O}$) of amide. ^1H NMR spectra of compounds shows presence singlet's at δ 3.88 (s, 2H, CH_2), δ 5.22 (s, 1H, CH) and δ 5.85 (s, 2H, NH_2 , D_2O exchangeable). MS (m/z): 295.37 (M^{+1}). (Found: C, 57.05; H, 3.29; N, 23.61; S, 10.65 require for $\text{C}_{14}\text{H}_9\text{N}_5\text{OS}$; C, 56.94; H, 3.07; N, 23.71; S, 10.86). Thus the data obtained confirmed conversion of *N*-acetyl chloride to 2-amino-1,3-thioazole is achieved. **3(a-b)** which were treated with methyl iodide in presence of triethyl amine to provide *N,N*-dimethyl aminothioazole compounds **4(a-b)**. IR spectra shows absence of strong band at 3512 cm^{-1} (NH_2) and ^1H NMR spectra shows singlet's at δ 2.78 (s, 6H, $\text{N}-\text{CH}_3$), δ 3.90 (s, 2H, CH_2), δ 5.45 (s, 1H, -CH) and absence of δ 5.85 (s, 2H, NH_2 , D_2O exchangeable). MS (m/z): 323.37 (M^{+1}). (Found: C, 59.33; H, 4.09; N, 21.51; S, 10.01

require for $\text{C}_{16}\text{H}_{15}\text{N}_5\text{OS}$: C, 59.43; H, 4.02; N, 21.66; S, 9.92). **4(a-b)** on further reaction with Hydrochloric acid gave amide compounds **5(a-b)**. IR spectra showed band at 3446 cm^{-1} (NH_2), 1759 cm^{-1} ($>\text{C}=\text{O}$) of ketone and 1640 cm^{-1} ($\text{HN}-\text{C}=\text{O}$) of amide and absence of band at 2241 cm^{-1} (CN). ^1H NMR spectra showed singlet for δ 3.28 (s, 3H, $\text{N}-\text{CH}_3$), δ 3.89 (s, 2H, CH_2), δ 4.20 (s, 2H, NH_2 , D_2O exchangeable), δ 4.60 (s, 1H, CH). MS (m/z): 341.39 (M^{+1}). (Found: C, 56.01; H, 4.39; N, 20.31; S, 9.28 and require for $\text{C}_{16}\text{H}_{15}\text{N}_5\text{O}_2\text{S}$: C, 56.29; H, 4.43; N, 20.51; S, 9.39). Thus from spectral data formation of amide compounds was confirmed. **4(a-b)** on further reaction with sodium azide and ammonium chloride in xylene gave tetrazole compounds **6(a-b)**. IR spectra shows band at 3365 cm^{-1} (NH), 1751 cm^{-1} ($>\text{C}=\text{O}$) of ketone and absence of band at 2241 cm^{-1} (CN). ^1H NMR spectra showed singlet for δ 3.21 (s, 3H, $\text{N}-\text{CH}_3$), δ 3.83 (s, 2H, CH_2), δ 5.19 (s, 1H, CH), δ 6.65-6.93 (m, 4H, Ar-H) and δ 8.48 (s, 1H, NH, D_2O exchangeable). MS (m/z): 366.40 (M^{+1}). (Found: C, 52.21; H, 3.96; N, 30.28; S, 8.68 require for $\text{C}_{16}\text{H}_{14}\text{N}_8\text{OS}$: C, 52.45; H, 3.85; N, 30.58; S, 8.75). Thus from spectral data formation of tetrazole compounds was confirmed.

The Inhibition zone against the bacterium *S. aureus*, *E. coli*, *S.typhi* and fungus *A. niger* due to the different substituted pyrrolobenzimidazole derivatives is shown in Table 1. Highest antimicrobial potential was observed with compounds 3b, 4b, 5b and 6b against *S. aureus*. Compounds 3a, 4a, 4b, 5a, 6b were found to be potent against *S.typhi* whereas compounds 2b, 4b, 5a, 6b exhibited inhibitory activity against *E. coli*. On the other hand compounds 2a, 2b, 3a, 3b, 5b and 6b showed highest antifungal potential against *A.niger*.

Table 1: Antibacterial activity and antifungal activity

Compounds	Zone of Inhibition (mm)							
	<i>S.aureus</i>		<i>S.typhi</i>		<i>E. coli</i>		<i>A. niger</i>	
	50 $\mu\text{g ml}^{-1}$	100 $\mu\text{g ml}^{-1}$	50 $\mu\text{g ml}^{-1}$	100 $\mu\text{g ml}^{-1}$	50 $\mu\text{g ml}^{-1}$	100 $\mu\text{g ml}^{-1}$	50 $\mu\text{g ml}^{-1}$	100 $\mu\text{g ml}^{-1}$
2a	11	17	12	18	13	15	16	20
2b	12	18	11	15	15	19	17	20
3a	14	18	15	20	10	15	15	21
3b	17	19	14	18	12	17	16	18
4a	15	20	15	21	14	16	12	17
4b	18	21	16	20	16	20	11	15
5a	12	18	15	19	15	20	10	16
5b	16	19	14	17	12	16	15	19
6a	14	18	12	18	13	17	11	16
6b	16	19	17	19	16	19	16	19

CONCLUSION

Spectral techniques used in the scheme, confirm the formation and synthetic route of novel pyrrolobenzimidazole derivatives. The results of antibacterial and antifungal activity depicts that synthesized derivatives exhibited significant to moderate activity. This confers that all the newly synthesized 2-aminothiazole, *N*, *N*-dimethyl aminothiazole, amide, tetrazole derivatives of pyrrolobenzimidazole are biologically active against the tested bacterial and fungal strains.

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REFERENCES

1. Preston PN. Synthesis, reactions, and spectroscopic properties of benzimidazoles. *Chem Rev.* 1974;74(3):279-314.
2. Vinodkumar R and Vaidya SD. Synthesis, antibacterial, antiasthmatic and antidiabetic activities of novel *N*-substituted-2-(4-phenylethynyl-phenyl)-1H-benzimidazoles and *N*-substituted-2[4-(4,4-dimethyl-thiochroman-6-yl-ethynyl)-phenyl]-1H-benzimidazoles. *Eur J Med Chem.* 2008;43(4):675- 986.
3. Yun H, Baogen W and Ranken R. 2-Piperidin-4-ylbenzimidazoles with broad spectrum antibacterial activities. *Bioorg Med Chem Lett.* 2003;13:3253.
4. Ozkay Y and Tunali Y. Antimicrobial activity and a SAR study of some novel benzimidazole derivatives bearing hydrazones moiety. *Eur J Med Chem.* 2010; 5(8):3293-8.
5. Singh N, Pandurangan A, Rana K and Anand P. Benzimidazole: A short review of their antimicrobial activities. *Int Curr Pharm J.* 2012;1(5):119-27.
6. Ayhan-Kilcigil G and Altanlar N. Synthesis and antimicrobial activities of some new benzimidazole derivatives. *Farmaco.* 2003;58:1345-1350.
7. Boiani M and Gonzalez M. Imidazole and benzimidazole derivatives as chemotherapeutic agents. *Mini-Rev Med Chem.* 2005;5:409-424.
8. Khabnadideh S, Rezaei Z and Ghafari N. Synthesis and antifungal activity of benzimidazole, benzotriazole and aminothiazole derivatives. *Res. Pharm Sci.* 2012;7(2):65-72.
9. Gallardo-Godoy A, Prusiner SB and Renslo AR. 2-Aminothiazoles as therapeutic leads for prion diseases. *J Med Chem.* 2011;54:1010-21.
10. Xie Y and Hui X. Synthesis and biological activity of amide compound containing pyrazole mandelic acid groups. *J Heterocyclic Chem.* 2017;54:2644.
11. Malath A, Al-Qatait and Haythem A. Synthesis, characterization, and biological activity of novel metronidazole-piperazine amides. *Monatshefte für Chemie. Chemical Monthly.* 2015;146(4):705-712.
12. Tang R and Jin L. Synthesis, antifungal and antibacterial activity for novel amide derivatives containing a triazole moiety. *Chemistry Central Journal.* 2013;7:30.
13. Adel AH and Abdel-Megeed. Synthesis and Antimicrobial Activity of New Tetrazoles Incorporating Isoindole-1,3-dione Moiety and Their Sugar Derivatives. *Journal of Heterocyclic Chemistry.* 2013;50(3):484-489.
14. Hutchinson DW and Naylor M. The antiviral activity of tetrazole phosphonic-acids and their analogs. *Nucleic Acids Research.* 1985;13(23):8519-8530.
15. Zafer A and Kaplancıklı L. Synthesis and antiproliferative activity of new 1,5-disubstituted tetrazoles bearing hydrazone moiety. *Med Chem Res.* 2014;23(2):1067-1075.
16. Maria M, Dorathi A and Jayanthi M. Synthesis, Characterization, Antibacterial & Anti- Inflammatory Effects Of Substituted Tetrazole Derivatives Based On Different Types Of Carbazone And Benzaldehyde. *International Journal of Chem Tech Research.* 2013;5(4):1982-1990.
17. Kiyoto K, Toshikazu O, Norihito T and Ryozo O. Process for preparation of 5-substituted tetrazoles. *US 5744612A.* 1998.