

EVALUATION OF IN VITRO ANTIOXIDANT ACTIVITY OF CALOTROPIS PROCERA FRUIT EXTRACT

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ABSTRACT

The plant *calotropis procera* (family-Asclepiadaceae) commonly known as " Aak " is used in many ayurvedic formulations. The present study was designed to evaluate the In vitro antioxidant potential of ethanolic fruit extract of *calotropis procera*. In vitro antioxidant activity was carried out by DPPH (1,1-Diphenyl-2-picryl hydrazyl) method, reducing power assay and phosphomolybdenum assay. The ethanolic fruit extract (100,250 and 500 µg/ml) was compared with Gallic acid (1, 2.5 and 5 µg/ml) as a standard. The results revealed that the fruit extract of *calotropis procera* possess significant antioxidant activity.

Keywords: *calotropis procera*, reducing power assay, phosphomolybdenum assay.

INTRODUCTION

Reactive oxygen species are involved in a number of degenerative diseases such as cancer, cirrhosis, atherosclerosis, diabetes, heart failure¹⁻⁴ and also in wound healing⁵. Plant-derived antioxidants such as tannins, stilbenes, phenolic acids, flavones could delay or provide protection for living organisms from damage caused by uncontrolled production of reactive oxygen species (ROS) and the concomitant lipid per oxidation, protein damage, DNA strand breaking⁶ because of their redox properties, which help them to act as hydrogen donors, reducing agents and free radical scavengers^{6, 7}. They are also strong chelators of metal ions⁸. Antioxidants play a potential role in the treatment and prevention of several diseases⁹. *Calotropis procera* (Ait.) R.Br. (giant milkweed) belongs to the family Asclepiadaceae locally known as "Aak" is distributed in tropical and sub-tropical Africa and Asia. *Calotropis procera* have large bushy shrub, leaves decussate, inflorescence extra axillary umbellate panicle, corolla purple, lobes erect¹⁰. The fruits are inflated, grey-green in colour and release flat,

brown seeds with a tuft of white hair at one end¹¹. The entire plant has been reported to contain alkaloids, sterols, flavonoids, cardiac glycosides, saponins, triterpenoids and uscharin¹².

Calotropis procera possess a wide range of biological activities such as analgesic, anti-inflammatory, antidiabetic, antiarthritic, antioxidant, antihelminthic, anticandidal, wound healing, anticonvulsant, antitumour, antiasthmatic and hepatoprotective.

MATERIALS AND METHODS

Collection of plant material

Fresh fruits were collected from the *Calotropis procera* plant growing in the surrounding areas of West Godavari district, A.P. The plant was identified and Authenticated by Dr. T. Raghuram, Taxonomist, Maharani's college, Peddapuram.

Preparation of extract

The freshly collected fruits of the plant were cleared from dirt and dried under shade and then coarsely powdered manually. The dried

fruit powder was macerated in ethanol for a period of 7 days and later subjected to hot percolation for 8 hrs. Then the solution was filtered, concentrated and subjected to drying.

Chemicals and Instruments

All the chemicals used are of Analytical grade. DPPH has been procured from Research lab fine chemical Industries, Mumbai, Gallic acid was a gifted sample. Trichloro acetic acid, sodium phosphate, ammonium molybdate and sulphuric acid were used for this study are of analytical grade. The Instruments used for this study are PH Meter, UV- Visible double beam spectrophotometer (ELICO-SL-210), centrifuge machine and electronic balance were used for the analysis.

IN-VITRO ANTIOXIDANT STUDY

DPPH Free Radical Scavenging Activity¹³

The free radical scavenging activity was evaluated by the DPPH method, 0.1mM solution of DPPH in ethanol was prepared, Gallic acid was taken as reference standard, different concentrations of the extract (100, 250 and 500 µg/ml) and standard drug (1, 2.5 and 5 µg/ml) were prepared using ethanol. 1.0 ml of 0.1 mM of DPPH solution was mixed with 3.0 ml of all the concentrations of extract and standard separately. A blank was prepared using 0.1 mM DPPH and ethanol mixture without adding extract. These mixtures are kept in dark about 30 min and the optical density was measured at 517 nm. The experiment was repeated triplicate. The percentage inhibition of the DPPH activity was calculated by using the following formula.

$$\text{DPPH Scavenged (\%)} = [(A_0 - A_1) / A_0] \times 100$$

Where A_0 is the absorbance of control reaction and A_1 is the absorbance of the sample extract. The antioxidant activity of the ethanolic fruit extract was expressed as IC_{50} and compared with standard. The IC_{50} Value was defined as the concentration (in µg / ml) of extract that Scavenges the DPPH radicals by 50%

Reducing power method¹⁴

Different concentrations of the extract (100,250 and 500 µg / ml) and standard drug (1, 2.5 and 5 µg/ml) were prepared using distilled water. 1% potassium ferricyanide, 10% Trichloroacetic acid, 0.1% ferric chloride and 0.2 M Phosphate buffer (pH 6.6) were prepared using distilled water. Gallic acid was taken as the reference standard. Then 1 ml of each concentration of extract and

standard were taken separately and mixed with 2.5 ml of 0.2 M phosphate buffer (pH 6.6) and 2.5 ml of potassium ferricyanide. Incubate all these samples at 50°C for 20 min. Then add 2.5 ml of 10% Trichloroacetic acid and centrifuge at 3000rpm for 10 min. now separate the upper layer (2.5ml) and then add (2.5ml) distilled water, 0.5 ml of freshly prepared ferric chloride. Then the absorbances were measured at 700nm.

Phosphomolybdenum Reduction Assay¹⁵

The antioxidant activity of the extract was evaluated by the phosphomolybdenum method. The assay is based on the reduction of MO (VI)-MO (V) by the extract and subsequent formation of green phosphate / MO (V) complex at acid pH. 0.3ml of each concentration of the extract and standard were taken separately and mixed with 3 ml of reagent solution (0.6M sulphuric acid, 28mM sodium phosphate and 4mM ammonium molybdate). The tubes containing the reaction solution were incubated at 95°C for 90 min. Then the absorbance of the solution was measured at 695nm using spectro photometer against blank after cooling to room temperature. Ethanol (0.3ml) in the place of extract was used as blank.

RESULTS AND DISCUSSION

DPPH Free Radical Scavenging Activity

The DPPH radical scavenging activity is a sensitive method for the antioxidant screening of plant extract¹⁶. The reduction capability of DPPH radicals was determined by decrease in its absorbance at 517nm, which is induced by antioxidants. **Table 1** shows the percentage of DPPH radical scavenged by Gallic acid and ethanolic fruit extract at various concentrations (µg / ml). **Figure 1a, 1b** illustrates a decrease in the concentration of DPPH radicals due to the scavenging ability of the soluble constituents in the ethanolic fruit extract of *Calotropis procera* and Gallic acid as a reference standard. The IC_{50} values were found to be 4.10 and 3.3 µg/ml for ethanolic fruit extract of *calotropis procera* and Gallic acid respectively.

Reducing Power Assay

Reducing power assay is based on the principle that substances, which have reduction potential, react with potassium ferricyanide (Fe^{+3}) to form potassium ferrocyanide (Fe^{+2}), which then reacts with ferric chloride to form ferrous complex that has an absorption maximum at 700nm. The reducing capacity of a compound may serve as

a significant indicator of its potential antioxidant activity. **Table 2** shows the reducing power of ethanolic fruit extract of *calotropis procera* and standard Gallic acid. From **Figure 2a, 2b** it was found that the absorbance of the extract increased with an increase in concentrations.

Phosphomolybdenum Reduction Assay

Phosphomolybdenum assay is based on the reduction of MO (VI) – MO (V) and forms green colour Phosphomolybdenum (V) complex, which shows maximum absorbance at 695nm. **Table 3** shows the antioxidant capacity of ethanolic fruit extract of *calotropis procera* and standard Gallic acid. From **Figure 3a, 3b** it was found that the absorbance of the extract increased with an increase in concentrations.

Table 1: DPPH Free Radical Scavenging Activity

Tested Material	Concentration (µg/ml)	% Inhibition ± SEM	IC ₅₀ µg/ml
Standard Gallic acid	1	24.32 ± 1.09	3.3 µg/ml
	2.5	40.68 ± 0.05	
	5	66.16 ± 0.42	
Ethanolic fruit extract of <i>calotropis procera</i>	100	67.59 ± 0.14	4.1 µg/ml
	250	75.92 ± 0.49	
	500	78.82 ± 0.53	

Table 2: Reducing power method

Tested Material	Concentration (µg/ml)	Absorbance ± SEM
Standard Gallic acid	1	0.3096 ± 0.0002
	2.5	0.3723 ± 0.0003
	5	0.5292 ± 0.0001
Ethanolic fruit extract of <i>calotropis procera</i>	100	0.086 ± 0.0001
	250	0.122 ± 0.0005
	500	0.275 ± 0.0005

Table 3: Phosphomolybdenum Assay

Tested material	Concentration (µg/ml)	Absorbance ± SEM
Standard Gallic acid	1	0.1626 ± 0.00003
	2.5	0.3149 ± 0.0015
	5	0.4291 ± 0.0003
Ethanolic fruit extract of <i>Calotropis procera</i>	100	0.1416 ± 0.0002
	250	0.2445 ± 0.0001
	500	0.3084 ± 0.0003

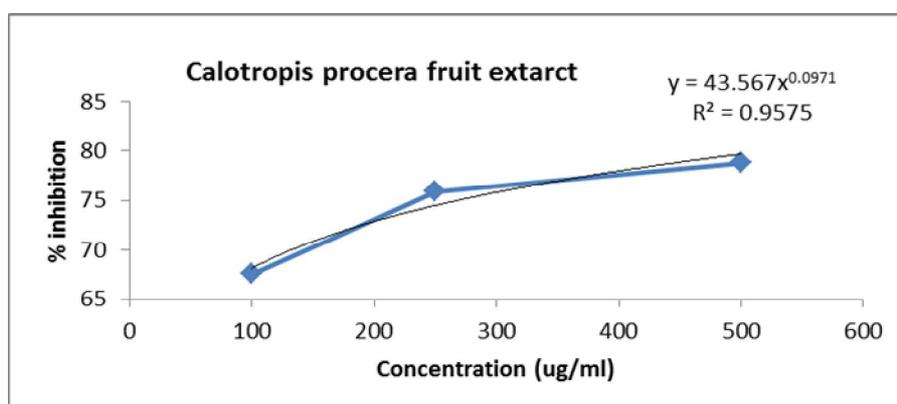


Fig. 1a: DPPH Free Radical Scavenging Activity of *Calotropis procera*

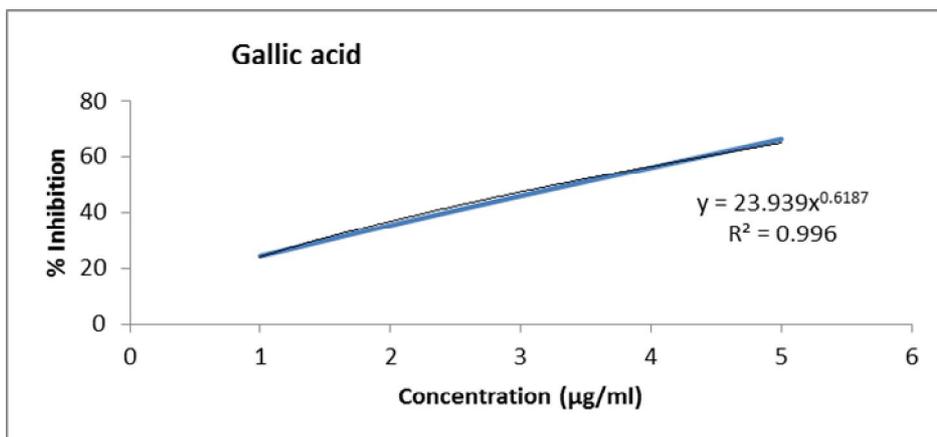


Fig. 1b: DPPH Free Radical Scavenging Activity of Gallic acid

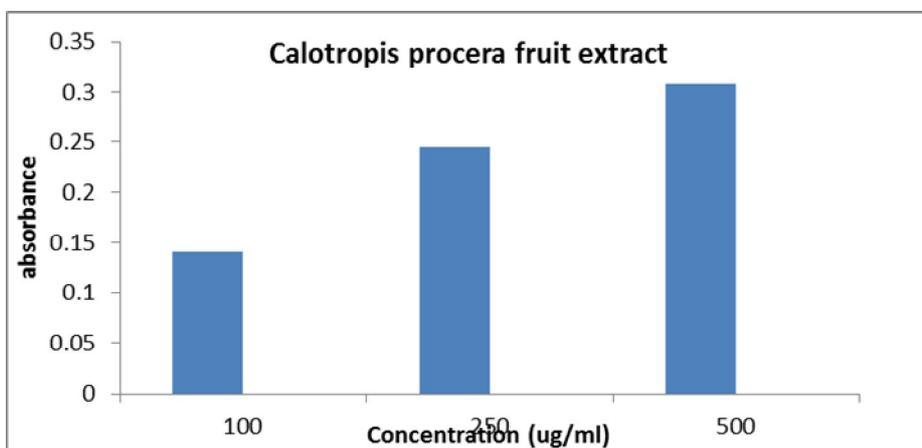


Fig. 2a: Reducing Power of *Calotropis procera*

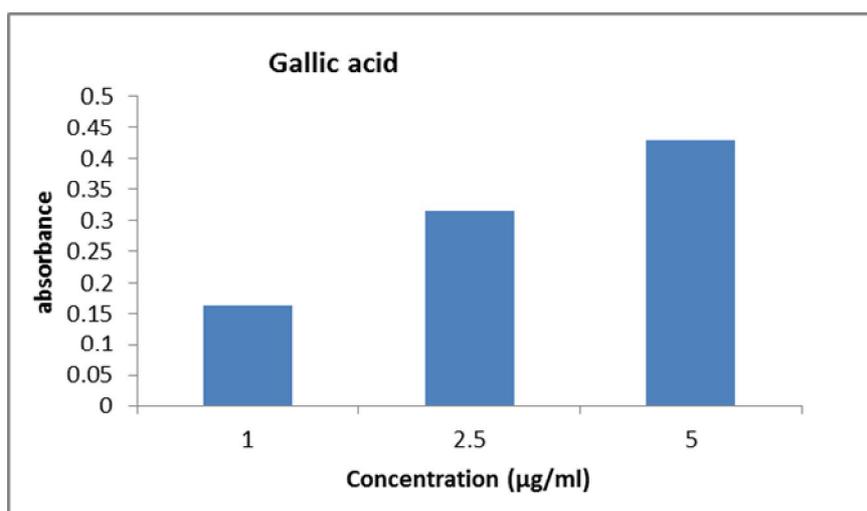


Fig. 2b: Reducing power of standard Gallic acid

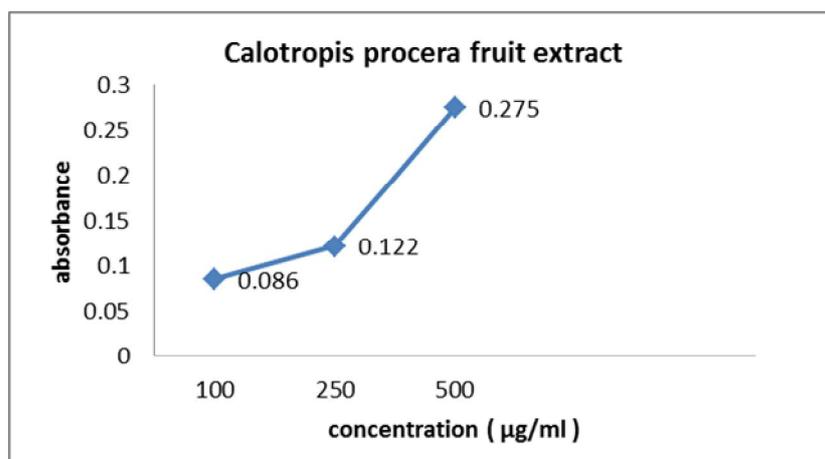


Fig. 3a: Phosphomolybdenum Antioxidant Assay of *Calotropis procera*

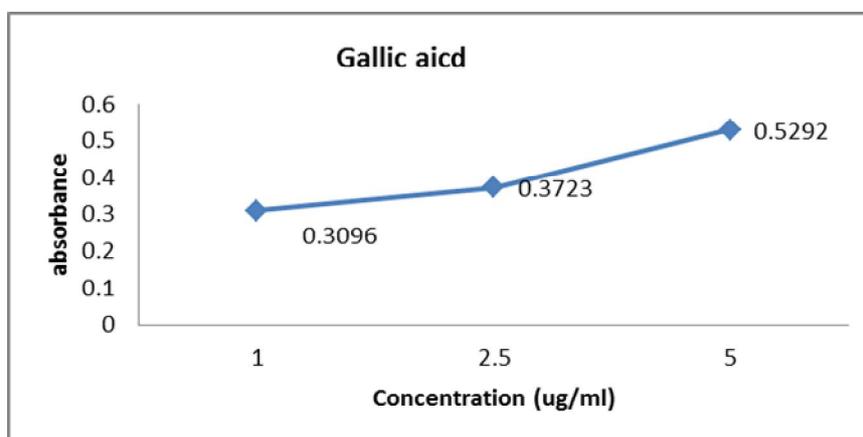


Fig. 3b: Phosphomolybdenum Antioxidant Assay of Gallic acid

CONCLUSION

The present study was designed to evaluate the in vitro antioxidant activity of ethanolic fruit extract of *calotropis procera*. The results obtained indicates the significant antioxidant activities in all the three methods and the results were compared with standard Gallic acid. Further work can be carried out to isolate the compounds and screen for their biological activities.

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